# A New Use of Cyclofunctionalisation With Selenenyl Reagents: An Example of Carbon-Carbon Bond Formation 

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Summary ( $Z, Z$ )-Cyclonona-1,5-diene reacts stereoselectively with PhSeCl in AcOH to give ( $2 ; \mathrm{R}=\mathrm{Ac}$ ).

Cyclofunctionalisation ${ }^{1}$ is an efficient route to heterocycles ${ }^{1,2}$ containing the synthetically useful benzeneselenogroup ${ }^{3}$ [equation (1)]. We have found a transannular example of this process that demonstrates how it can be used to make a $\mathrm{C}-\mathrm{C}$ bond [equation (1), $\mathrm{X}=\mathrm{C}] . \dagger$


Under appropriate conditions the readily available ${ }^{4}$ diene (1) reacts with PhSeCl to give the substituted hydrindan (2; $R=A c$ ). ${ }^{5}$ Thus, when an AcOH solution ( 0.04 m ) of $\mathrm{PhSeCl}\left(1.02\right.$ equiv.) is added ( 0.3 equiv. $\mathrm{min}^{-1}$ ) to the diene in $\mathrm{AcOH}(0.04 \mathrm{~m})$ containing anhydrous NaOAc ( 1 equiv.) the liquid hydrindanyl acetate ( $\mathbf{2} ; \mathrm{R}=\mathrm{Ac}$ ) is produced. $\ddagger$ This compound, isolated 10 min after the end of the addition is treated with $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}-\mathrm{K}_{2} \mathrm{CO}_{3}(5 \mathrm{ml}, 0.85 \mathrm{ml}, 84 \mathrm{mg}$, respectively, per mmol of acetate) at room temperature for 16 h and the crystalline alcohol ( $2 ; \mathrm{R}=\mathrm{H}$ ) can be isolated as an analytically pure solid (m.p. $85-86^{\circ} \mathrm{C}$ ) in $68 \%$ yield [based on (1)]. Acetylation $\left(\mathrm{Ac}_{2} \mathrm{O}\right.$, pyridine, 3.5 h , room
temp.) of ( $2 ; \mathrm{R}=\mathrm{H}$ ) regenerates ( $70 \%$ ) the origina reaction product ( $\mathbf{2} ; \mathrm{R}=\mathrm{Ac}$ ). Treatment of the alcohol (2; $\mathrm{R}=\mathrm{H}$ ) with $\mathrm{Ph}_{3} \mathrm{SnH}^{6}$ in refluxing PhMe gives the alcohol (3) $(90 \%)$ shown, by comparison with an authentic sample ${ }^{7}$ and by comparison of the derived sharp-melting acid phthalates, ${ }^{7}$ to have the structure and stereochemistry defined by (3).§


(2)

On the basis of the firmly established ${ }^{8}$ trans mode of addition of selenenyl reagents to double bonds the transannular reactions ${ }^{9}$ of (1) are expected to proceed as in (4) or (5) T to place the $\mathrm{PhSe}-$ and AcO - groups in a cis-1,4-relationship and trans to the bridgehead hydrogens. The validity of these predictions, and the preferential formation of (6) [ $=(\mathbf{2})$ ] rather than (7), is established by the reduction to (3) together with the following observations.

The n.m.r. signal $\left(\mathrm{CDCl}_{3} ; 100 \mathrm{MHz}\right)$ centred at $\delta 5.0$ due to $\mathrm{H}_{\mathrm{a}}$ [see (6)] is a $1: 2: 2: 2: 1$ quintet corresponding to a pair of overlapping triplets ( $J_{\mathrm{ab}}=J_{\mathrm{ac}}=5 \cdot 2 ; J_{\mathrm{ad}}=12 \cdot 1$ Hz ). The signal centred at $\delta 3.58$ due to $\mathrm{H}_{\mathrm{e}}$ is a sextet $\left(J_{\mathrm{ef}}=c a . \quad J_{\mathrm{eh}}=9.9 ; J_{\mathrm{eg}}=5 \cdot 8 \mathrm{~Hz}^{10}\right) . \quad$ Dreiding models
$\dagger$ For a previous attempt to make a $\mathrm{C}-\mathrm{C}$ bond using PhSeCl , see ref. 1.
$\ddagger$ With EtOAc as solvent (and no NaOAc), a mixture of ( $2 ; R=A c$ ) and ( $2 ; R=H$ ) is obtained.
§ Compound (3), and not the C-4 epimer, is the main product obtained by oxymercuration-reduction of (1) (cf. G. Nagendrappa and D. Devaprabhakara, Tetrahedron Letters, 1970, 4687; S. N. Moorthy and D. Devaprabhakara, ibid., 1975, 257).

If For simplicity, other conformations of (7) are not shown. The pathway via (5) involves a boat transition state and has not been observed in other (ref. 9 and the ref. in footnote §) transannular reactions of (1).

(4)
(6)


(5)
reveal that the observed pair of splitting patterns due to $\mathrm{H}_{\mathrm{a}}$ and $\mathrm{H}_{\mathrm{e}}$ would not arise from (7). Therefore, as anticipated on mechanistic grounds, ${ }^{9}$ the $\mathrm{PhSe}-$ group is attached to the $\mathrm{C}_{5}$-ring in the orientation shown. This assignment is corroborated by mass spectral measurements. A diagnostically significant peak ( $21 \%$ of base peak) occurs at $m / e$ 97.0652 in the spectrum of (3) and corresponds to $\left[\mathrm{C}_{6} \mathrm{H}_{9} \mathrm{O}\right]^{+}$. Such a species is expected in terms of one of the standard fragmentation pathways of cyclohexanols ${ }^{11}$ [see (8); $\mathrm{R}=\mathrm{H}]$. When (2) is reduced with $\mathrm{Ph}_{3} \mathrm{SnD}^{6}$ the alcohol produced should, and does, give a corresponding peak for $\left[\mathrm{C}_{6} \mathrm{H}_{8} \mathrm{DO}\right]^{+}(8 ; \mathrm{R}=\mathrm{D})(13.5 \%$ of base peak, $m / e$ found 98.0714 , calc. $98 \cdot 0716$ ).

(8)

Correct mass measurements and satisfactory ( $\pm 0.3$ ) combustion analytical data were obtained for all new compounds.

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